

C208—N203—C202—O201	-177.9 (3)
C204—N203—C202—O215	-177.5 (3)
C208—N203—C202—O215	0.3 (5)
C202—N203—C204—C205	99.5 (4)
C208—N203—C204—C205	-78.5 (4)
C204—N203—C208—C209	171.9 (3)
C202—N203—C208—C209	-5.9 (5)
N203—C204—C205—N206	166.0 (3)
N203—C204—C205—O207	-17.6 (5)
N203—C208—C209—C210	-175.4 (3)
N203—C208—C209—C214	4.5 (4)
C208—C209—C210—C211	-179.3 (3)
C214—C209—C210—C211	0.8 (4)
C208—C209—C214—O215	2.1 (4)
C208—C209—C214—C213	-179.3 (3)
C210—C209—C214—O215	-178.0 (2)
C210—C209—C214—C213	0.6 (4)
C209—C210—C211—C212	-1.6 (5)
C210—C211—C212—C213	1.1 (5)
C211—C212—C213—C214	0.3 (6)
C212—C213—C214—O215	177.5 (3)
C212—C213—C214—C209	-1.1 (5)

Table 3. Hydrogen-bonding geometry (Å, °)

D—H...A	H...A	D—H...A
N106—H161...O107 ⁱ	2.104 (48)	166.4 (45)
N106—H162...O101 ⁱⁱ	2.162 (46)	167.0 (37)
N206—H261...O207 ⁱⁱⁱ	2.032 (48)	166.2 (37)
N206—H262...O201 ^{iv}	1.978 (44)	163.0 (35)

Symmetry codes: (i) $x + \frac{1}{2}, 1 - y, z$; (ii) $-x, 1 - y, z + \frac{1}{2}$; (iii) $x + \frac{1}{2}, 2 - y, z$; (iv) $-x, 2 - y, z + \frac{1}{2}$.

The structure was solved using *SHELXS86* (Sheldrick, 1990) and refined with *SHELXL93* (Sheldrick, 1993). Most of the calculations were performed with the *SHELXL93* package, as was the generation of the Crystallographic Information File (CIF) used for the submission of this paper.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, complete geometry and torsion angles have been deposited with the IUCr (Reference: DU1075). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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3-[5-(2-Nitrophenyl)furfurylidene]-2,4-pentanedione

V. VRÁBEL

*Department of Analytical Chemistry,
Faculty of Chemical Technology,
Slovak Technical University, Radlinského 9,
812 37 Bratislava, Slovakia*

J. LOKAJ

*Central Laboratory of Chemical Technics,
Faculty of Chemical Technology,
Slovak Technical University, Radlinského 9,
812 37 Bratislava, Slovakia*

J. SIVÝ

*Department of Analytical Chemistry,
Faculty of Pharmacy, Comenius University,
880 37 Bratislava, Slovakia*

D. ILAVSKÝ AND A. BARTOVIČ

*Department of Organic Chemistry,
Faculty of Chemical Technology,
Slovak Technical University, Radlinského 9,
812 37 Bratislava, Slovakia*

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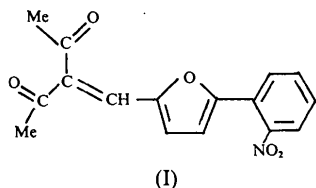
Abstract

The title compound, $C_{16}H_{13}NO_5$, consists of monomeric units. The whole molecule was tested for planarity. Unlike the 3-nitrophenyl isomer [Lokaj, Vrabel, Sivý, Ilavský & Koreňová (1994). *Acta Cryst.* **C50**, 1312–1314], the fragment containing the O(1), C(1), C(2) and C(5) atoms, the fururylidene ring and the attached nitrophenyl group are not in the same plane. The dihedral angle between the benzene and furan rings is 47.8°.

Comment

The conformation of furan derivatives with an *R* substituent at position 5 and a $—CH=CR'_2$ substituent at position 2 can be assumed on the basis of previously reported information (Kusá, Polynova, Poray-Koshits, Kováč & Végh, 1979; Kusá, Polynova, Poray-Koshits & Juráček, 1979; Kusá, Polynova, Poray-Koshits & Kováč, 1979). The conformation depends on the type of *R'* substituent; electron-acceptor substituents form *trans* and

electron-donor substituents form *cis* conformations with respect to the double bond between the furan and ethylene fragments. In the present molecule, (I), one can predict a *trans* conformation of the vinyl-furan fragment on the basis of the *R'* substituent furan. This was confirmed by the X-ray analysis.



The starting compound (2-nitrophenyl)furaldehyde was prepared by a published procedure (Krutošiková, Kováč, Frimm, Kováč & Sticzay, 1971). 5-(2-Nitrophenyl)-2-furaldehyde (0.02 mol) was dissolved in tetrahydrofuran (25 ml) and then 2,4-pentanedione (0.02 mol) was added. The reaction was catalyzed by anhydrous CuCl₂ and the product was extracted by chloroform (3 × 60 ml) and evaporated to dryness. The crude product was recrystallized from ethanol. The molecular structure and the atom numbering of the title compound are shown in Fig. 1.

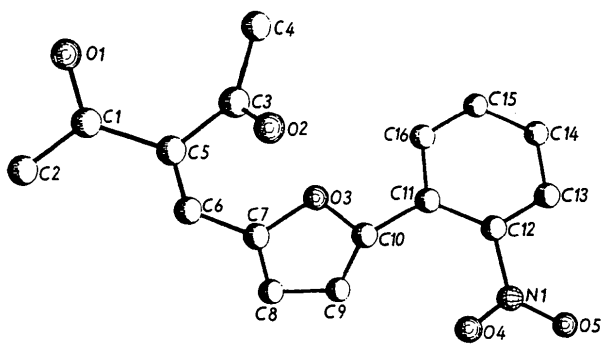


Fig. 1. Perspective drawing of the title compound with atom numbering.

Unlike the 3-nitrophenyl isomer (Lokaj, Vrabel, Sivý, Ilavský & Koreňová, 1994) where the O(1)—C(1)—C(2)—C(5) fragment, the furan ring and the nitrophenyl group lie nearly in the same plane, resulting in significant conjugation through the whole molecule, no such planarity was observed in the present molecule. Inserting the nitro group at the *ortho* position of the phenyl ring results in a twisting of the phenyl ring against the furan ring. The angle between the best planes of these two rings is 47.8 (1)°. There is a conjugation along the

O(1)=C(1)—C(5)=C(6)—C(7)=C(8) fragment, as indicated by the planarity of this moiety as well as by the corresponding bond distances. Comparing the C(1)—C(5), C(6)—C(7) and C(7)=C(8) bond lengths with the data (using the 3 σ criterion) reported by Allen, Kennard, Watson, Brammer & Orpen (1987), it was found that the C(1)—C(5) bond length agrees well with the C_{sp³}—C_{sp³} bond length of Allen *et al.* (1987) [C=C—C(=O)—C 1.464 Å]. The C(6)—C(7) bond length is shortened, whereas the C(7)=C(8) and C(9)=C(10) double-bond lengths agree excellently with the value of 1.341 Å reported by Allen *et al.* (1987) for the C_{sp²}=C_{sp²} distance in the furan system. The phenyl ring is approximately planar [maximum deviation of atoms C(11) and C(12) from the least-squares plane is 0.005 (2) Å with an average C—C bond distance of 1.376 (3) Å].

Experimental

Crystal data

C₁₆H₁₃NO₅

M_r = 299.282

Triclinic

P $\bar{1}$

a = 8.243 (10) Å

b = 8.717 (9) Å

c = 10.652 (9) Å

α = 67.91 (7)°

β = 81.73 (9)°

γ = 74.09 (9)°

V = 681 (1) Å³

Z = 2

D_x = 1.46 Mg m⁻³

D_m = 1.43 (2) Mg m⁻³

D_m measured by flotation in ZnSO₄ solution

Mo K α radiation

λ = 0.71069 Å

Cell parameters from 15 reflections

θ = 4–17°

μ = 0.099 mm⁻¹

T = 293 K

Prism

0.25 × 0.20 × 0.10 mm

Orange

Crystal source: crystallization from CCl₄

Data collection

XP2₁ diffractometer

$\theta/2\theta$ scans

Absorption correction:

none

3285 measured reflections

3153 independent reflections

1641 observed reflections

[*I* > 1.96 σ (*I*)]

R_{int} = 0.026

θ_{\max} = 27.5°

h = -10 → 10

k = -10 → 10

l = 0 → 13

2 standard reflections

monitored every 98

reflections

intensity variation: 5%

Refinement

Refinement on *F*

R = 0.054

wR = 0.051

S = 0.84

1641 reflections

199 parameters

H-atom parameters not

refined

Unit weights applied

(Δ/σ)_{max} = 0.1

$\Delta\rho_{\max}$ = 0.23 e Å⁻³

$\Delta\rho_{\min}$ = -0.27 e Å⁻³

Atomic scattering factors

from *International Tables*

for X-ray Crystallography

(1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)
$$B_{\text{eq}} = (4/3)\sum_i \sum_j \beta_{ij} a_i \cdot a_j$$

	x	y	z	B_{eq}
O(1)	0.6350 (2)	-0.3605 (2)	0.8980 (2)	4.10 (5)
O(2)	0.4832 (2)	0.0647 (2)	0.7128 (2)	4.23 (6)
O(3)	0.1272 (2)	0.1550 (2)	0.8365 (1)	2.74 (4)
O(4)	0.0337 (2)	0.6269 (2)	0.7431 (2)	4.64 (6)
O(5)	-0.1667 (3)	0.8214 (2)	0.6337 (2)	6.39 (8)
N(1)	-0.0891 (2)	0.6736 (2)	0.6770 (2)	3.80 (6)
C(1)	0.5480 (2)	-0.2948 (2)	0.9744 (2)	2.95 (7)
C(2)	0.5679 (3)	-0.3749 (3)	1.1233 (2)	3.53 (7)
C(3)	0.4229 (3)	-0.0553 (2)	0.7683 (2)	3.01 (7)
C(4)	0.3476 (4)	-0.1297 (3)	0.6925 (2)	4.86 (10)
C(5)	0.4205 (2)	-0.1344 (2)	0.9191 (2)	2.56 (6)
C(6)	0.3175 (2)	-0.0639 (2)	1.0004 (2)	2.72 (6)
C(7)	0.1810 (2)	0.0842 (2)	0.9662 (2)	2.72 (6)
C(8)	0.0908 (3)	0.1765 (3)	1.0404 (2)	3.02 (7)
C(9)	-0.0244 (3)	0.3122 (3)	0.9549 (2)	3.18 (7)
C(10)	0.0011 (2)	0.2954 (2)	0.8332 (2)	2.69 (6)
C(11)	-0.0902 (2)	0.3763 (3)	0.7085 (2)	2.79 (6)
C(12)	-0.1458 (3)	0.5509 (3)	0.6423 (2)	3.16 (6)
C(13)	-0.2463 (3)	0.6169 (3)	0.5327 (2)	4.03 (8)
C(14)	-0.2909 (3)	0.5085 (4)	0.4860 (2)	4.71 (9)
C(15)	-0.2373 (3)	0.3361 (4)	0.5476 (2)	4.47 (9)
C(16)	-0.1374 (3)	0.2711 (3)	0.6574 (2)	3.66 (8)

Table 2. Selected bond distances (\AA) and angles ($^\circ$)

O(1)—C(1)	1.209 (3)	N(1)—C(12)	1.456 (3)
O(2)—C(3)	1.195 (3)	C(1)—C(2)	1.487 (3)
O(3)—C(7)	1.369 (2)	C(1)—C(5)	1.469 (3)
O(3)—C(10)	1.364 (3)	C(3)—C(4)	1.489 (4)
O(4)—N(1)	1.204 (3)	C(3)—C(5)	1.490 (3)
O(5)—N(1)	1.212 (3)	C(10)—C(11)	1.455 (3)
C(7)—O(3)—C(10)	105.7 (1)	O(1)—C(1)—C(2)	120.8 (2)
O(4)—N(1)—O(5)	122.6 (2)	C(2)—C(1)—C(5)	119.7 (2)
O(5)—N(1)—C(12)	117.5 (2)	O(2)—C(3)—C(5)	119.6 (2)
O(4)—N(1)—C(12)	119.8 (2)	O(2)—C(3)—C(4)	122.4 (2)

Data reduction was performed with the program *XP2₁* (Pavelčík, 1987). The structure was solved by direct methods and refinement was by block-diagonal least squares. H atoms were found from difference Fourier maps and their coordinates and displacement parameters were fixed. Calculations were performed with a local version of the *NRC* system (Ahmed, Hall, Pippy & Huber, 1973) and *PARST* (Nardelli, 1984).

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: KA1054). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Enantioselective Synthesis of Isoquinoline Alkaloids from Simple Sugar. II. Structure of 1-(6,7-Dimethoxy-3,4-dihydroisoquinolin-1-yl)-1,2,3,4-butanetetrayl Tetraacetate *N*-Oxide

ZBIGNIEW CZARNOCKI

University of Warsaw, Department of Chemistry, 02-093 Warsaw, Poland

JAN K. MAURIN

Institute of Atomic Energy, Solid State Physics Department, 05-400 Otwock-Świerk, Poland

MAŁGORZATA WINNICKA-MAURIN

Agricultural University of Warsaw, Institute of Chemistry, Warsaw, Poland

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Abstract

The title compound, $\text{C}_{23}\text{H}_{29}\text{NO}_{11}$, is an important intermediate product in the course of the synthesis of some natural products. Among them are two isoquinoline alkaloids: (*R*)-calycotomine and (*S*)-xylopinine. A single-crystal X-ray study of this compound is reported.

Comment

In the course of our work on the development of an efficient enantioselective synthetic strategy for isoquinoline alkaloids, we described a procedure in which the chiral substrate *D*-ribonic acid γ -lactone (1) was used (Czarnocki, 1992*a,b*). In several steps, compound (1) was transformed into 6,7-dimethoxy-3,4-dihydroisoquinolin-1-yl-(1',2',3',4'-tetraacetoxo)butane (2) (Czarnocki, 1992*a,b*, and references therein). Subsequent oxidation of (2) with *m*-chloroperbenzoic acid gave 1-(6,7-dimethoxy-1,2,3,4-tetrahydro-1,2-epoxyisoquinolinyl)-1,2,3,4-butanetetrayl tetraacetate (3)